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APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. H0002067 USA (4030) 1624 Troy W. Francisco 01/18/2002 10/051,512 **EXAMINER** 05/20/2004 7590 NGUYEN, NGOC YEN M Colleen D. Szuch, Esq; Honeywell Law Dept. Honeywell International, Inc. ART UNIT PAPER NUMBER 101 ColumbiaRoad, Building Meyer 5 Morristown, NJ 07962 1754

DATE MAILED: 05/20/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

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		Applicati	on No.	Applicant(s)	
OFF. 1.4.		10/051,5	10/051,512 FRANCISCO ET AL.		
	Office Action Summary	Examine	r	Art Unit	
			n M. Nguyen	1754	
Period fo	The MAILING DATE of this communication or Reply	n appears on the	e cover sheet with the	e correspondence addres	SS
THE - Exte after - If the - If NC - Failt Any	ORTENED STATUTORY PERIOD FOR R MAILING DATE OF THIS COMMUNICATION sions of time may be available under the provisions of 37 Cl SIX (6) MONTHS from the mailing date of this communication period for reply specified above is less than thirty (30) days, period for reply is specified above, the maximum statutory pure to reply within the set or extended period for reply will, by reply received by the Office later than three months after the ed patent term adjustment. See 37 CFR 1.704(b).	ON. FR 1.136(a). In no evon. a reply within the stateriod will apply and wastatute, cause the app	ent, however, may a reply be lutory minimum of thirty (30) o iill expire SIX (6) MONTHS fr blication to become ABANDO	timely filed days will be considered timely om the mailing date of this commu NED (35 U.S.C. § 133).	nication.
Status					
	Responsive to communication(s) filed on <u>13 August 2003</u> . This action is FINAL . 2b) This action is non-final. Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.				
Disposit	ion of Claims				
5)⊠ 6)⊠	Claim(s) 1-12 and 14-19 is/are pending in 4a) Of the above claim(s) is/are with Claim(s) 11,12,14,18 and 19 is/are allowe Claim(s) 1-3,5-10,15 and 16 is/are rejecte Claim(s) 4, 17 is/are objected to. Claim(s) are subject to restriction a	hdrawn from co d. d.	nsideration.		
Applicat	ion Papers				
10)	The specification is objected to by the Exa The drawing(s) filed on is/are: a) Applicant may not request that any objection to Replacement drawing sheet(s) including the co The oath or declaration is objected to by the	accepted or b) the drawing(s) borrection is require	pe held in abeyance. Sometimed if the drawing(s) is a	See 37 CFR 1.85(a). objected to. See 37 CFR 1.	` ,
Priority (ınder 35 U.S.C. § 119				
a)	Acknowledgment is made of a claim for for All b) Some * c) None of: 1. Certified copies of the priority docur 2. Certified copies of the priority docur 3. Copies of the certified copies of the application from the International Business the attached detailed Office action for a	ments have bee ments have bee priority docume ureau (PCT Rul	en received. en received in Applica ents have been recei e 17.2(a)).	ation No ived in this National Staç	ge
Attachmen	• •		_		
2) 🔲 Notic 3) 🔲 Infori	e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTO-948 nation Disclosure Statement(s) (PTO-1449 or PTO/SI r No(s)/Mail Date		4) Interview Summa Paper No(s)/Mail 5) Notice of Informa 6) Other:)

Art Unit: 1754

DETAILED ACTION

Claims 11-12, 14, 18-19 are allowed.

Claims 4, 17 are objected to as being dependent upon a rejected base claim, but would be allowable if rewritten in independent form including all of the limitations of the base claim and any intervening claims.

The following is a statement of reasons for the indication of allowable subject matter: the prior art does not teach or suggest the use of an infrared analyzer equipped with a diamond-tipped probe to determine the concentration of free and bound water in an acid stream.

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-3, 5-10, 15-16 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 97/35187.

WO discloses a process for continuous production of hydrogen fluoride comprises (1) a step of reacting starting fluorspar with starting sulfuric acid, (2) a step of separating a crude reaction product as obtained in step (1) into (a) a low-boiling mixture composed predominantly of hydrogen

Art Unit: 1754

fluoride and (b) a high-boiling mixture composed predominantly of unreacted sulfuric acid and containing small proportions of hydrogen fluoride and water,

(3) a step of purifying and isolating hydrogen fluoride from said low-boiling mixture (a), (4) a step of adding sulfuric anhydride to said high-boiling mixture (b) in a substantially equivalent amount with respect to the water to convert substantially all the water to sulfuric acid and returning it together with said unreacted sulfuric acid as sulfuric acid feed to said step (1), and (5) a step of adding sulfuric acid in a supplemental amount to provide the amount of sulfuric acid needed for reaction with starting fluorspar, wherein, in said step (4), the amount of water occurring in said high-boiling mixture (b) is determined by the method of the invention for determining the component concentration, particularly the concentration of water, of a ternary mixture (note page 4, lines 6-page 6, 31 and page 7, second full paragraph).

The process for determining the concentration of water is described in WO '187 as a method of determining the concentration of each component of a ternary mixture essentially consisting of sulfuric acid, hydrogen fluoride, and water, which comprises measuring at least one set of the three physical quantities, namely (1) temperature, (2) ultrasound propagation velocity, and (3) electrical conductivity or viscosity, of the ternary mixture and converting measured values to the concentrations of the respective components according to calibration curves representing the relationships of the concentrations of respective components of a ternary mixture composed of sulfuric acid, hydrogen

Art Unit: 1754

fluoride, and water with the above-mentioned three physical quantities as separately constructed beforehand (note paragraph bridging pages 6-7).

WO '187 further discloses that the use of a computer is desired in order to conduct the treatment accurately and fast (note first paragraph on page 14).

The difference is WO '187 does not specifically disclose the use of a probe in at least a portion of the feed stream.

It would have been obvious to one of ordinary skill in the art at the time of the invention was made to use any known means in the art to performed the functions required in WO '187 in order to accurately estimate the amount of water, without a showing of criticality or unexpected results, the use such probe is not seen as a patentable difference since the probe is a known and commercially available means in the art.

Applicant's arguments filed August 13, 2003 have been fully considered but they are not persuasive.

Applicants argue that WO '187 fails to teach or suggest a method of regulating water content including the step of estimating the amount of bound and free water in a stream. This is fundamentally different that a method which measures just the total concentration of water in the stream.

It appears that in there are only two types of water in a mixture, bound (i.e., chemically bound) and free water. Therefore, there is no difference seen between the claimed "bound and free water" and the total concentration of water as disclosed in the

Art Unit: 1754

applied reference. In the event that there is a difference, it should be noted that Applicants' claim 1 only require to "estimate", thus, the total concentration of water in the applied reference is a good estimate for the "bound and free water".

Applicants argue that WO '187 fails to tech or suggest any method for regulating and measuring water in a stream comprising sulfuric acid, hydrofluoric acid and fluorosulfonic acid, and water.

In WO '187, even though only sulfuric acid, hydrogen fluoride (i.e. hydrofluoric acid), and water are disclosed, not fluorosulfonic acid, however, since WO '187 discloses that the hydrogen fluoride is produced by reacting fluorspar and sulfuric, some fluorosulfonic acid may also be present in the mixture (note Applicants' specification, page 1, lines 22-29, which state that when fluorspar and acid are used to produce HF, an acid mixture is produced, which frequently comprises "a major proportion of sulfuric acid and minor proportions hydrofluoric acid and fluorosulfonic acid"). Furthermore, Applicants' claims do not require any particular amount for the fluorosulfonic acid, it can be present in the mixture in impurity amount.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the

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Art Unit: 1754

shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Ngoc-Yen M. Nguyen whose telephone number is (571) 272-1356. The examiner is currently on Part time schedule.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Stanley Silverman can be reached on (571) 272-1358. The fax phone numbers for the organization where this application or proceeding is assigned are (703) 872-9306 for regular communications and (703) 872-9306 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is (703) 308-0661.

Ngoc-Yen M. Nguyen Primary Examiner

Art Unit 1754

nmn

May 13, 2004